



Chemical reaction

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Chemical reaction is a process in which one or more substances are converted into one or more new substances. This process involves the breaking and forming of chemical bonds between atoms. Chemical reactions are often represented by chemical equations, which show the reactants on the left and the products on the right, separated by an arrow. The law of conservation of mass states that the total mass of the reactants must equal the total mass of the products.

Chemical reactions can be classified into several types, including synthesis, decomposition, single displacement, double displacement, and combustion. Each type of reaction has its own characteristic features and examples. For example, synthesis reactions involve the combination of two or more substances to form a single product, while decomposition reactions involve the breakdown of a single substance into two or more products.

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Libros y documentos

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FECHA DE PUBLICACIÓN

1995

FORMATO

Artículo

DATOS DE PUBLICACIÓN

Cuando un amigo se va [artículo] Milton Aguilar.

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